- (1) Benzenesulfonamide, 4-amino-N-(aminoiminomethy1)- (sulfaguanidine); C₇H₁₀N₄O₂S; [57-67-0]
- (2) Water; H₂O; [7732-18-5]

ORIGINAL MEASUREMENTS:

Roblin, R. O., Jr.; Williams, J. H.; Winnek, P. S.; English, J. P. J. Am. Chem. Soc. 1940, 62, 2002-5.

VARIABLES:

One temperature: 37°C

PREPARED BY:

R. Piekos

EXPERIMENTAL VALUES:

Solubility of sulfaguanidine in water at 37° C is 190 mg/100 cm³ solution (8.87 x 10^{-3} mol dm⁻³, compiler).

AUXILIARY INFORMATION

METHOD/APPARATUS/PROCEDURE:

Excess sulfonamide in water was heated and stirred on a steam bath for 30 min. The suspension was then agitated for 24 h in a thermostat at 37°C. A sample of the satd soln was withdrawn through a glass filter, dild, and analyzed by the Marshall method (1) using a General Electric recording spectrophotometer for comparing the colors developed with those of the standards.

SOURCE AND PURITY OF MATERIALS:

Sulfaguanidine, mp 189-90°C (dec, cor) was prepd by the authors. Anal: %C 39.2 (calcd 39.3); %H 4.6 (4.7); %N 21.7 (22.4).

Purity of the water was not specified.

ESTIMATED ERROR:

Nothing specified.

REFERENCES:

 Bratton, A. C.; Marshall, E. K., Jr. J. Pharmacol. <u>1939</u>, 66, 4.

- (1) Benzenesulfonamide, 4-amino-N-(aminoiminomethyl)- (sulfaguanidine); $^{1}_{7}^{1}_{10}^{1}_{4}^{0}_{2}^{0}_{5}$; [57-67-0]
- (2) Water; H₂O; [7732-18-5]

ORIGINAL MEASUREMENTS:

Sapozhnikova, N. V.; Postovskii, I. Ya. Zh. Prikl. Khim. 1944, 17, 427-34.

VARIABLES:

Temperature

PREPARED BY:

R. Piekos

EXPERIMENTAL VALUES:

	Solubility			
t/°C	Weight%	10 ² mol kg ⁻¹ water ^a		
20	0.760	3.58		
37	0.196	.92		
50	0.430	2.02		
75	1.40	6.63		
99	3.70	17.93		

a Calculated by compiler.

AUXILIARY INFORMATION

METHOD/APPARATUS/PROCEDURE:

Sulfaguanidine was dissolved in water to form a satd soln which was occasionally agitated in a glass vessel immersed in a thermostat. The equilibrium was usually attained after 1 h. Five- to 100-cm³ samples of the satd soln were placed in Pt crucibles or dishes and evapd to dryness at temps lower than 110-115°C. The residue was dried to const wt at 105-110°C and weighed.

SOURCE AND PURITY OF MATERIALS:

Pure, recrystd sulfaguanidine was used. Its mp conformed to that reported in the literature.

Purity of the water was not specified.

ESTIMATED ERROR:

Soly: quite reliable results were obtained over the temp range 20-75°C. At higher temps the accuracy was poor due to evaph of water during sampling (authors).

Temp: +0.05°C (authors).

REFERENCES:

COMPONENTS: (1) Benzenesulfonamide, 4-amino-N-(amino-iminomethyl)- (sulfaguanidine); C7H10N4O2S; [57-67-0] (2) Water; H2O; [7732-18-5] VARIABLES: One temperature: 18-19°C EXPERIMENTAL VALUES: ORIGINAL MEASUREMENTS: Becher, R.; Leya, S. Experientia 1946, 2, 459-60.

Solubility of sulfaguanidine in water at room temperature (18-19 $^{\circ}$ C) is 65 mg% (3.0 x 10 $^{-3}$ mol dm $^{-3}$).

AUXILIARY INFORMATION

After standing for more than two days the soln of sulfaguanidine in water was filtered and sulfaguanidine was assayed in

METHOD/APPARATUS/PROCEDURE:

the filtrate colorimetrically by the method of Druey and Oesterheld (1).

SOURCE AND PURITY OF MATERIALS: Nothing specified.

ESTIMATED ERROR:

Nothing specified.

REFERENCES:

 Druey, J.; Oesterheld, G. Helv. Chim. Acta 1942, 25, 753.

- (1) Benzenesulfonamide, 4-amino-N-(aminoiminomethy1)- (sulfaguanidine); C₇H₁₀N₄O₂S; [57-67-0]
- (2) Water; H₂O; [7732-18-5]

ORIGINAL MEASUREMENTS:

Gerencsér-Németh, M.; Horváth, M. Gyógyszerészet 1973, 17, 417-21.

VARIABLES:

One temperature: 20°C

PREPARED BY:

R. Piekos

EXPERIMENTAL VALUES:

Solubility of sulfaguanidine in water at $20^{\circ}\mathrm{C}$ is 0.1111 g/100 g solution $(5.191 \times 10^{-3} \text{ mol kg}^{-1} \text{ water, compiler}).$

AUXILIARY INFORMATION

METHOD/APPARATUS/PROCEDURE:

A weighed escess of sulfaguanidine in water was shaken in a shaker at 120 rpm for 6 h. The soln was then filtered, the residue was washed with the filtrate and finally with a small amt of distd water, dried and weighed.

SOURCE AND PURITY OF MATERIALS:

Sulfaguanidine (source not specified) was dried at 100°C for 3 h or over conc H2SO, for 72 h. Its mp was 187.5-8.8°C.

Distd water was used.

ESTIMATED ERROR:

Soly: precision \pm 0.0047 g/100 g (2 detns)

(compiler)

Temp; not specified.

AUXILIARY INFORMATION

METHOD/APPARATUS/PROCEDURE:

After standing for more than two days the soln of sulfaguanidine was filtered and sulfaguanidine was assayed in the filtrate colorimetrically by the method of Druey and Oesterheld (1).

SOURCE AND PURITY OF MATERIALS:

Nothing specified.

ESTIMATED ERROR:

Nothing specified.

REFERENCES:

 Druey, J.; Oesterheld, G. Helv. Chim. Acta 1942, 25, 753.

COMPONENTS: (1) Benzenesulfonamide, 4-amino-N-(amino-iminomethyl)- (sulfaguanidine); C7H10N4O2S; [57-67-0] (2) Sodium chloride, NaCl; [7647-14-5] (3) Water; H2O; [7732-18-5] VARIABLES: Temperature ORIGINAL MEASUREMENTS: Avico, U.; Cavazutti, G.; di Francesco, R.; Signoretti Ciranni, E.; Zuccaro, P. Farmaco, Ed. Pratica 1975, 30(1), 40-6.

EXPERIMENTAL VALUES:

Solubility of amorphous sulfaguanidine in equimolal NaCl solutions

t/°C	equ1mo	equimolal NaCl solutions		
	g/100 g water	10 ³ mol kg ⁻¹ water ^a		
25	0.71	3.3		
35	0.84	3.9		
40	0.93	4.3		

a Calculated by compiler.

AUXILIARY INFORMATION

METHOD/APPARATUS/PROCEDURE: A soln of sulfaguanidine-HCl was added to an NaOH soln contg stoichiometric quantity of the base to neutralize the HCl salt. The neutralization was carried out in a thermostat and the pH of the mixt was maintained close to that of a satd soln of sulfaguanidine in water. The procedure was repeated using various initial concns of the reagents to find the max concn of sulfaguanidine at which no pptn occurred.

SOURCE AND PURITY OF MATERIALS:

Source and purity of sulfaguanidine was not specified. The mp of crystalline sulfaguanidine was $190-3^{\circ}C$.

Purity of the water was not specified.

ΞS	TIMA	TED	ERR	OR:

Nothing specified.

- (1) Benzenesulfonamide, 4-amino-N-(aminoiminomethyl)- (sulfaguanidine); C₇H₁₀N₄O₂S; [57-67-0]
- (2) Pectin; $(C_{13}^{H}_{18}^{O}_{12})_{n}$; [9000-69-5]
- (3) Water; H₂O; [7732-18-5]

ORIGINAL MEASUREMENTS:

Becher, R.; Leya, S., Experientia 1946, 2, 459-60.

VARIABLES:

One temperature: 18-19°C

PREPARED BY:

R. Piekos

EXPERIMENTAL VALUES:

Solubility of sulfaguanidine in a 2.5% pectin solution ([pectin] = 6.8×10^{-2} mol kg⁻¹, compiler), of pH about 2.6, at room temperature (18-19°C) is 111 mg% (5.18 $\times 10^{-3}$ mol dm⁻³, compiler).

AUXILIARY INFORMATION

METHOD/APPARATUS/PROCEDURE:

The soln was allowed to stand for more than 2 days at room temp. The soln was then filtered, and sulfaguanidine assayed in the filtrate colorimetrically by the method of Druey and Oesterheld (1).

SOURCE AND PURITY OF MATERIALS:

A high quality apple pectin was used: the rel viscosity of a 0.5% soln was 6.2, and for neutralization of 1 g of the pectin, 1.67 cm³ of a 1 mol dm⁻³ NaOH soln was used. The source and purity of sulfaguanidine and water were not specified.

ESTIMATED ERROR:

Nothing specified.

REFERENCES:

 Druey, J.; Oesterheld, G., Helv. Chim. Acta 1942, 25, 753.

COMPONENTS: (1) Benzenesulfonamide, 4-amino-N-(amino-iminomethyl)- (sulfaguanidine); C₇H₁₀N₄O₂S; [57-67-0] (2) Pectinic acid, sodium salt; (C₁₃H₁₇NaO₁₂)_n; [9049-37-0] (3) Water; H₂O; [7732-18-5] VARIABLES: One temperature: 18-19°C ORIGINAL MEASUREMENTS: Becher, R.; Leya, S., Experientia 1946, 2, 459-60.

EXPERIMENTAL VALUES:

Solubility of sulfaguanidine in a 2.6% neutral sodium pectinate solution ([sodium pectinate] = $6.7 \times 10^{-2} \text{ mol kg}^{-1}$ (n = 1), compiler) at room temperature (18-19°C) is 101 mg% (4.71 x 10^{-3} mol dm⁻³, compiler).

AUXILIARY INFORMATION

METHOD/APPARATUS/PROCEDURE:

The soln was allowed to stand for more than two days at room temp. The soln was then filtered, and sulfaguanidine assayed in the filtrate colorimetrically by the method of Druey and Oesterheld (1).

SOURCE AND PURITY OF MATERIALS:

Nothing specified.

ESTIMATED ERROR:

Nothing specified.

REFERENCES:

1. Druey, J.; Oesterheld, G., Helv. Chim. Acta 1942, 25, 753.

- (1) Benzenesulfonamide, 4-amino-N-(aminoiminomethyl)- (sulfaguanidine); C₇H₁₀N₄O₂S; [57-67-0]
- (2) Ethanol; C₂H₆O; [64-17-5]
- (3) Water; H₂O; [7732-18-5]

ORIGINAL MEASUREMENTS:

Sapozhnikova, N. V.; Postovskii, I. Ya. Zh. Prikl. Khim. 1944, 17, 427-34.

VARIABLES:

Concentration of ethanol

PREPARED BY:

R. Piekos

EXPERIMENTAL VALUES:

Concentration of ethanol	Solubility at 75°C			
Weight%	Weight%	mol kg ⁻¹ solvent ^a		
0	1.40	0.0663		
19.2	2.67	0.128		
57.6	4.43	0.216		
76.4	5.56	0.275		
86	4.80	0.235		
96	3.53	0.171		

a Calculated by compiler.

AUXILIARY INFORMATION

METHOD/APPARATUS/PROCEDURE:

Sulfaguanidine was dissolved in EtOH-water mixtures to form satd solns which were occasionally agitated in glass vessels immersed in a thermostat. The equilibrium was usually attained after 1 h. Five- to 100-cm³ samples of the satd soln were placed in Pt crucibles or dishes and evapd to dryness at temps lower than 110-115°C. The residue was dried to const at 105-110°C and weighed.

SOURCE AND PURITY OF MATERIALS:

Pure, recrystd sulfaguanidine was used. Its mp conformed with that reported in the literature.

The purity of ethanol and water was not specified.

ESTIMATED ERROR:

Soly: quite reliable results were obtained (authors).

Temp: ±0.05°C (authors).

COMPON	FNTS •	ORIGINAL MEASUREMENTS:
(1)	Benzenesulfonamide, 4-amino-N-(amino-iminomethyl)- (sulfaguanidine); $C_7^H_{10}^N_4^0_2^S$; [57-67-0] Ethanol; $C_2^H_6^0$; [64-17-5] 1,2,3-Propanetriol; $C_3^H_8^0$; [56-81-5]	Dolique, R.; Foucault, J. Trav. soc. pharm. Montpellier 1952, 12, 145-53.
	Water; H ₂ O; [7732-18-5]	
VARIAB		PREPARED BY:
	One temperature: 26-28°C	R. Piekos
EXPERI	MENTAL VALUES:	
	Solubility of sulfaguanidine in a mixt	ure of 1,2,3-propanetriol and 95° ethanol
	(2:1 by wt) at 26-28°C is 4% (0.2 mol	kg solvent, compiler).
		INFORMATION
метної	AUXILIARY D/APPARATUS/PROCEDURE:	INFORMATION SOURCE AND PURITY OF MATERIALS:
The	D/APPARATUS/PROCEDURE:	SOURCE AND PURITY OF MATERIALS:
The	D/APPARATUS/PROCEDURE: sulfaguanidine content was detd by	SOURCE AND PURITY OF MATERIALS:
The diaz	D/APPARATUS/PROCEDURE: sulfaguanidine content was detd by zotization of the amine group in a cold dified 0.1N KNO ₂ soln. An excess of	SOURCE AND PURITY OF MATERIALS:
The diaz acid KNO	D/APPARATUS/PROCEDURE: sulfaguanidine content was detd by zotization of the amine group in a cold diffied 0.1N KNO ₂ soln. An excess of was detected by using iodinated	SOURCE AND PURITY OF MATERIALS:
The diaz	D/APPARATUS/PROCEDURE: sulfaguanidine content was detd by zotization of the amine group in a cold diffied 0.1N KNO ₂ soln. An excess of was detected by using iodinated	SOURCE AND PURITY OF MATERIALS:
The diaz acid KNO	D/APPARATUS/PROCEDURE: sulfaguanidine content was detd by zotization of the amine group in a cold diffied 0.1N KNO ₂ soln. An excess of was detected by using iodinated	SOURCE AND PURITY OF MATERIALS:
The diaz acid KNO	D/APPARATUS/PROCEDURE: sulfaguanidine content was detd by zotization of the amine group in a cold diffied 0.1N KNO ₂ soln. An excess of was detected by using iodinated	SOURCE AND PURITY OF MATERIALS:
The diaz acid KNO	D/APPARATUS/PROCEDURE: sulfaguanidine content was detd by zotization of the amine group in a cold diffied 0.1N KNO ₂ soln. An excess of was detected by using iodinated	SOURCE AND PURITY OF MATERIALS:
The diaz acid KNO	D/APPARATUS/PROCEDURE: sulfaguanidine content was detd by zotization of the amine group in a cold diffied 0.1N KNO ₂ soln. An excess of was detected by using iodinated	SOURCE AND PURITY OF MATERIALS: Nothing specified.

313 COMPONENTS: ORIGINAL MEASUREMENTS: (1) Benzenesulfonamide, 4-amino-N-(amino-Dolique, R.; Foucault, J. iminomethyl)-:(sulfaguanidine); Trav. soc. pharm. Montpellier 1952, 12, $C_7H_{10}N_4O_2S; [57-67-0]$ (2) Ethanol; $C_2H_6O; [64-17-5]$ 145-53. (3) 1,2,3-Propanetriol; C₃H₈O₃; [56-81-5] (4) Urea; CH₄N₂O; [57-13-6] (5) Water; H₂O; [7732-18-5] VARIABLES: PREPARED BY: One temperature: 26-28°C R. Piekos EXPERIMENTAL VALUES: Solubility of sulfaguanidine at 26-28°C in a saturated solution of urea in a mixture of 1,2,3-propanetriol and 95° ethanol (2:1 by wt), containing 54.5g of urea per 100 g of the mixture, is 5.77% (0.286 mol kg⁻¹ solvent, compiler). AUXILIARY INFORMATION METHOD/APPARATUS/PROCEDURE: SOURCE AND PURITY OF MATERIALS: The sulfaguanidine content was detd by Nothing specified. diazotization of the amine group in a cold acidified 0.1N KNO, soln. An excess of KNO2 was detected by using iodinated starch. ESTIMATED ERROR: Nothing specified. REFERENCES:

COMPONENTS: ORIGINAL MEASUREMENTS: Benzenesulfonamide, 4-amino-N-(amino-Becher, R.; Leya, S. Experientia 1946, 2, (1) iminomethyl)- (sulfaguanidine) C₇H₁₀N₄O₂S; [57-67-0] 459-60. (2) D-Glucose; C₆H₁₂O₆; [50-99-7] (3) Water; H₂O; [7732-18-5] VARIABLES: PREPARED BY: One temperature: 18-19 °C R. Piekos EXPERIMENTAL VALUES: Solubility of sulfaguanidine in a 10% D-glucose solution at room temperature $(18-19^{\circ}C)$ is 70 mg% $(3.3 \times 10^{-3} \text{ mol dm}^{-3}, \text{ compiler})$. AUXILIARY INFORMATION METHOD/APPARATUS/PROCEDURE: SOURCE AND PURITY OF MATERIALS: After standing for more than two days the Nothing specified. soln of sulfaguanidine was filtered and the sulfonamide was assayed in the filtrate colorimetrically by the method of Druey and Oesterheld (1). ESTIMATED ERROR: Nothing specified. REFERENCES: 1. Druey, J.; Oesterheld, G. Helv. Chim. Acta 1942, 25, 753.

- (1) Benzenesulfonamide, 4-amino-N-(amino-iminomethyl)- (sulfaguanidine); ${}^{C}_{7}{}^{H}_{10}{}^{N}_{4}{}^{O}_{2}{}^{S};$ [57-67-0]
- (2) Sorbitan monooleate, polyoxyethylene derivatives (Tween 80);[9005-65-6]
- (3) Water; H₂O; [7732-18-5]

ORIGINAL MEASUREMENTS:

Gerencsér-Németh, M.; Horváth, M. Gyógyszerészet 1973, 17, 417-21.

VARIABLES:

Concentration of Tween 80

PREPARED BY:

R. Piekos

EXPERIMENTAL VALUES:

Concentration of Tween 80	Solubility at 20°C			
Weight%	g/100 g soln ^a	10 ³ mol kg ⁻¹ soln ^b		
1	0.1131 0.1137	5.279 5.307		
3	0.1449 0.1471	6.763 6.866		
5	0.1636 0.1633	7.636 7.622		
8	0.2078 0.2090	9.699 9.755		

^a Numerical values supplied by the authors.

AUXILIARY INFORMATION

METHOD/APPARATUS/PROCEDURE:

An excess of sulfaguanidine in an aq Tween 80 soln was shaken in a lab shaker at 120 rpm for 6 h. The soln was then filtered, the residue was washed first with the filtrate and finally with a small amt of water, dried and weighed.

SOURCE AND PURITY OF MATERIALS:

Sulfaguanidine (sour ∞ not specified) was dried at 100° C for 3 h or over conc H_2SO_4 for 72 h. Its mp was $187.5-8.8^{\circ}$ C.

Source and purity of Tween 80 was not specified.

Distd water was used.

ESTIMATED ERROR:

Nothing specified.

b Calculated by compiler.

- (1) Benzenesulfonamide, 4-amino-N-(aminoiminomethyl)- (sulfaguanidine); C7H10N4O2S; [57-67-0]
- (2) 2-Propanol; C₃H₈O; [67-63-0]

ORIGINAL MEASUREMENTS:

Burlage, H. M.

J. Am. Pharm. Assoc., Sci. Ed. <u>1948</u>, 37, 345.

VARIABLES:

One temperature: 25°C

PREPARED BY:

R. Piekos

EXPERIMENTAL VALUES:

Solubility of sulfaguanidine in 2-propanol at 25° C is 0.1770 g/100 cm³ solution (8.262 x 10^{-3} mol dm⁻³, compiler).

AUXILIARY INFORMATION

METHOD/APPARATUS/PROCEDURE:

Satd solns of sulfaguanidine in 2-propanol were prepd at 25°C and definite vols of the solns were measured into tared dishes by means of standard pipets. The alcohol was allowed to evap at room temp and the residue was dried at 105°C. In the case of losses due to apparent decompn, the residue was dried in a desiccator (1).

SOURCE AND PURITY OF MATERIALS:

The sulfaguanidine was manufd by Squibb and was of the U.S.P. purity.

The source and purity of 2-propanol was not specified.

ESTIMATED ERROR:

Nothing specified.

REFERENCES:

 Burlage, H. M. J. Am. Pharm. Assoc., Sci. Ed. <u>1947</u>, 36(1), 16.

- (1) Benzenesulfonamide, 4-amino-N-(aminoiminomethyl)- (sulfaguanidine); $C_7H_{10}N_4O_2S$; [57-67-0]
- (2) 2-Propanone (acetone); C₃H₆O; [67-64-1]

ORIGINAL MEASUREMENTS:

Gutierrez, F. H.

Anales fis. quim. (Madrid) 1945, 41, 537-60.

VARIABLES:

Temperature

PREPARED BY:

R. Piekos

EXPERIMENTAL VALUES:

t/°C	· · · · · · · · · · · · · · · · · · ·						
	G ^a	Ep	X _g /1 ^c	mol/1 ^d acetone	mmol/mol acetone	1:X _g e	1 + X _{cc} f
0	1.548	1.524	12,610	58.8	4.2	64.60	79.30
5	1.574	1.539	12.730	59.4	4.3	63.53	78.55
10	1.622	1.596	13.025	60.7	4.4	62.89	76.75
15	1.679	1.651	13.383	62.4	4.5	59.56	74.74
20	1.728	1.700	13.672	63.8	4.7	57.87	73.15
25	2.013	1.973	15.806	73.8	5.4	49.73	63.25
30	2.215	2.177	17.261	80.6	6.0	45.15	57.94
35	2.584	2.519	19.982	93.3	7.0	38.70	50.05
40	2.929	2.846	22.847	104.9	7.9	35.14	44.48
45	3.199	3.099	24.357	113.7	8.7	31.26	41.05
50	3.549	3.427	26.813	125.1	9.6	28.18	37.29

 $[\]frac{a}{G} = \frac{p}{P} \frac{100}{p}$, where p and P are the weights of solute and solution, resp.

AUXILIARY INFORMATION

METHOD/APPARATUS/PROCEDURE:

A special all-glass app was constructed enabling the prepn of satd solns, agitation by bubbling a stream of acetone-satd N, filtration, and distn off the solvent without contact with air. Two exchangeable dissoln vessels of 15 and 8 cm³ working capacity were used depending on the soly of solute. The app was immersed in a thermostat. The vols of acetone used were 15 or 5 cm³, and the equilibration time was 2-2.5 h. The satd solns were filtered, weighed, the solvent was distd off, the residues were dried at 105°C, weighed, and examd for the presence of solvated acetone.

SOURCE AND PURITY OF MATERIALS:

The source of the materials was not specified. Pure, anhyd acetone was used. The absence of impurities and water was confirmed by procedures of the German Pharmacopeia VI and Spanish Pharmacopeia VIII.

The purity of sulfaguanidine was not specified.

ESTIMATED ERROR:

Soly: measurements were repeated until 2 values not differing in the second decimal were obtained (author).

Temp: ±0.1°C (author).

b E = $\frac{G\ 100}{B\ +\ 100}$; c g/1 acetone; d should be mmo1/1 acetone (compiler);

e g of acetone require to dissolve 1 g of solute; f volume (cm³) of acetone required to dissolve 1 g of solute.